

## Study on accumulation of copper and cadmium metals in *Gracilariaverrucosa* in nutrition enriching condition and their effect on growth of the organism

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**Abstract:** Pollution being the measure threat in this developing world, different methodologies are being invented or discovered every year. Algae, sea weeds, aquatic plants have been identified as potential resources for accumulating and bioconcentrating heavy metals. In this study a test is formulated how nutrient enrichment enhances the metal tolerance of floating macrophytes taking *Gracilariaverrucosa* as the test organism. Relative growth rates were measured in *Gracilariaverrucosa* exposed to different cadmium and copper concentrations in laboratory conditions. Photosynthetic pigment levels were negatively correlated with metal exposures, and nutrient addition attenuated chlorophyll decrease in response to metal exposures. MDA content and EC also showed sharp increases at higher concentrations, indicating oxidative stress. Relative growth rates were observed with metal exposure, it is found that the addition of nutrients helped the growth of the test organism, which often suppressed in the concentrated heavy metal exposure. This study indicates that nutrient enrichment increases the tolerance of *Gracilariaverrucosa* to metals, which is a suitable organism for controlling low level of Cadmium and copper in waste water.

**Keywords:** *Gracilariaverrucosa*, heavy metals, nutrient, toxic effect, biological activity

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### I. Introduction

Water contamination with heavy metals has become a severe problem in industrialized areas. The presence of toxic heavy metals in ditch, lake and river water badly affects the aquatic life and well-being of people who depend upon these sources for their daily water requirements. Consumption of aquatic food stuffs contaminated with toxic metals may cause serious health hazards through bio-magnification (Wenhua et al., 2007). A number of methods are available to remove toxic metals from water, including ion exchange, reverse osmosis, electrolysis, precipitation and adsorption which are very expensive. The latter is by far the most versatile and widely used. However, these methods have different efficiencies for different metals and can be prohibitively expensive, especially in large volumes, at low metal concentrations and in the presence of biota and suspended solids. The choice of purification method depends on the composition of the system, the pH of the water, redox conditions, and the nature of the pollutants. It is well known that aquatic plants can accumulate metals taken in from the environment and concentrate them in trophic chains with accumulative effect (Wenhua, H., Chen, X., Song, G., Wang, Q. and Chang, C.C. (2007)). Significant copper accumulation has been observed in other aquatic plants such as *Ceratophyllum demersum*, *Hydrilla verticillata*, *Lemna trisulca*, *Vallisneria spiralis* and *Lemna minor* (Razinger et al., 2009). The use of aquatic macrophytes, such as water hyacinth, duckweed, and water lettuce, in wastewater treatment has attracted global attention in recent years (Mohan, B.S. and Hosetti, B.B. (1997), as these plants can be grown on the surface of stabilization ponds, and may contribute to nutrient recovery from wastewater. Duckweed is a floating aquatic macrophyte belonging to the botanical family Lemnaceae, which can be found world-wide on the surface of nutrient-rich fresh and brackish waters. Mohan, B.S. and Hosetti, B.B. (1997) reported that an increase in nitrate concentration resulted in a significant increase in cadmium accumulation in *Ulva fasciata*. According to Padhi et al., 2010, nutrient enrichment either attenuated (chromium and zinc) or suppressed (nickel) root biomass decrease in response to metal exposure in *Salvinia herzogii*. Although heavy metal accumulation has been studied in *L. gibba* (Demirezen et al., 2007), the effect of nutrient addition on this species' metal tolerance is unknown. The aim of this study was to determine whether nutrient enrichment enhances the metal tolerance of floating macrophytes, which would therefore enable the growth of floating vegetation in constructed wetlands at metal concentrations that would otherwise inhibit plant viability.

## II. Materials And Methods

### 2.1 Gracilariaverrucosa (Hudson) Papenfuss

Gracilariaverrucosa was collected from Kalijai area of Chilika Lake. Plants variable in length, about 15-35 cm in length attach to small stones and often worm tubes in bottom. (Padhi et al., 2010) Thallus erect, terete, highly branched, and attached by a small circular disc found with a single main axis, uniformly thin throughout branching lateral, occasional sub-dichotomous, opposite or sub-opposite in arrangement, frequently alternate branch tips attenuated, branches up to the third or even fourth order, ultimate branches 4-12 cm in length and 0.5 mm in thickness. Cortex of two to four layers of cells. Outermost radially elongated  $10\ \mu\text{m} \times 7\ \mu\text{m}$ . Some being modified to form enlarged cyst like cells produced on the surface as tubular hairs, inner cortical cell slightly larger and more or less oval. Medulla of 3-5 layers of cells around a large spherical or ovoid central cell about  $300\ \mu\text{m}$  across cells of medulla progressively smaller from the center outwards, oval  $18\ \mu\text{m} \times 13\ \mu\text{m} - 14\ \mu\text{m} \times 15\ \mu\text{m}$ . (Panigrahy, H.B, 2013). Transition in size from medulla to cortex is abrupt, Spermatangia in deep cavities. Carpogonial branching system with a supporting cell bearing a two celled carpogonial branch and two sterile branches of one or two cells each. Cystocarps sub-spherical elevated and scattered,  $700\ \mu\text{m}-900\ \mu\text{m}$  in height,  $750-850\ \mu\text{m}$  wide at the base and  $900-1050\ \mu\text{m}$  wide at the broadest part. Gonimoblast of densely aggregated radiating filaments. Carpospores in a short series,  $10-13\ \mu\text{m}$  in diameter sympodially developed at ends of gonimoblast filaments. Gonimonemata radiating. Tetrasporangia scattered  $28-35\ \mu\text{m}$  long and  $16-22\ \mu\text{m}$  wide. (Swain, P. K., 2009)

### 2.2 Treatment condition:

The chemical composition of the lake water was (mean  $\pm$  standard deviation): pH=6.5 $\pm$ 0.2; conductivity =  $92 \pm 8\ \mu\text{Scm}^{-1}$ ;  $\text{SO}_4^{2-} = 0.32 \pm 0.02\ \text{mg L}^{-1}$ ;  $\text{NH}_4^+-\text{N} = 0.023 \pm 0.01\ \text{mg L}^{-1}$ ;  $\text{NO}_3^--\text{N} = 0.022 \pm 0.001\ \text{mg L}^{-1}$ ;  $\text{NO}_2^--\text{N} = 0.001 \pm 0.001\ \text{mg L}^{-1}$ . Before metal treatment, plants were acclimatized for 5 days in laboratory conditions in lake water ( $23^\circ\text{C}$  and 14h photoperiod,  $350\ \mu\text{mol m}^{-2}\ \text{s}^{-1}$ ). Plants were later transferred to deionized water with nutrient and different metal concentrations. Plants were treated with different concentrations of copper (0, 1, 2, 5, 10  $\text{mg L}^{-1}$ ) and cadmium (0, 0.5, 1, 2, 4  $\text{mg L}^{-1}$ ) and maintained in double deionized water in 500 mL conical flasks under the aforementioned conditions for periods of 1, 3, 5 and 7 days. They were grown in deionized water added with the nutrients in concentration that mimicked natural pond water. Maximum concentrations of copper and cadmium were taken from Wenhua (2007). Plant growth rates in response to metal exposures were compared with exposures enriched with 5  $\text{mg L}^{-1}$  P ( $\text{KH}_2\text{PO}_4$ ), 5  $\text{mg L}^{-1}$   $\text{NO}_3^--\text{N}$  ( $\text{KNO}_3$ ) and  $\text{SO}_4^{2-}$  ( $\text{K}_2\text{SO}_4$ ). Flasks without metals grown alongside each set of experimental groups served as controls. The experiments lasted 7 days, when the plants exposed to the highest metal concentrations developed chlorosis and necrosis. After harvesting, plants were washed with double deionized water. Plants were placed on blotting paper and allowed to drain for 5 min before weighing. All treatments were carried out in triplicate. Gracilariaverrucosa relative growth rates were calculated in each group according to Hunt's equation:

$$R = \frac{\ln W_2 - \ln W_1}{T_2 - T_1}$$

Where, R is the relative growth rate ( $\text{gg}^{-1}\text{d}^{-1}$ ),  $W_1$  and  $W_2$  are the initial and final fresh weights, respectively, and ( $T_2 - T_1$ ) is the experimental period (Hunt, 1978).

### 2.3 Quantification of heavy metals:

After harvesting the plants were washed thoroughly with double deionized water, blotted and oven dried at  $85^\circ\text{C}$ . Each sample was then digested with 10 mL pure  $\text{HNO}_3$ , using a CEM-MARS 5 (CEM Corporation Mathews, NC, USA) microwave digestion system (maximum power: 1,200 W, power: 100%, ramp: 20:00 min, pressure: 180 psi, temperature:  $210^\circ\text{C}$  and hold time: 10:00 min). After digestion, the volume of each sample was adjusted to 25 mL using double deionized water. Determinations of Cd and Cu concentrations in all samples were carried out by inductively coupled plasma optical emission spectroscopy (Varian-Liberty II, ICP-OES) (Demirezen, 2007). Peach leaves (NIST, SRM-1547) and CRM 039-050 were used as reference material and also all analytical procedures were performed for reference materials. Samples were analyzed in triplicate.

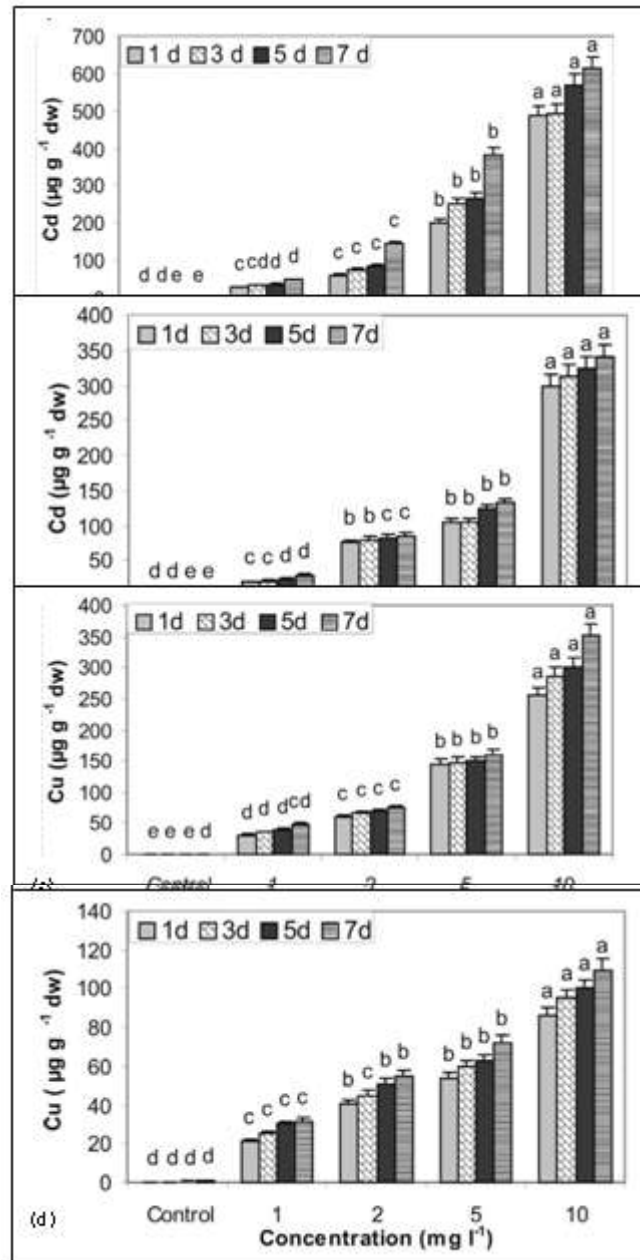
### 2.3 Plant growth parameters

Plant biomass was measured on the basis of fresh weight. Photosynthetic pigments of treated and untreated plants (100 mg) were extracted in 80% chilled acetone in dark. After centrifugation at 10,000 g for 10 min, absorbance of the supernatant was taken at 450, 645 and 663 nm. The content of chlorophylls and carotenoids were estimated by as previously described (Witham, 1971).

### 2.4 Lipid peroxidation and electrical conductivity

Lipid peroxidation was determined by estimation of the malondialdehyde (MDA) content following the protocol of Health and Packer (1968). MDA and EC content were not determined for the treatments. Influence of nutrient addition on growth and accumulation of cadmium and copper in

*Gracilariaverrucosa* enriched with nutrients. Plant material (500 mg) was homogenized with 3 mL of 0.5% thiobarbituric acid in 20% trichloroacetic acid. The homogenate was incubated at 95°C for 30 min and reactions were stopped on ice.



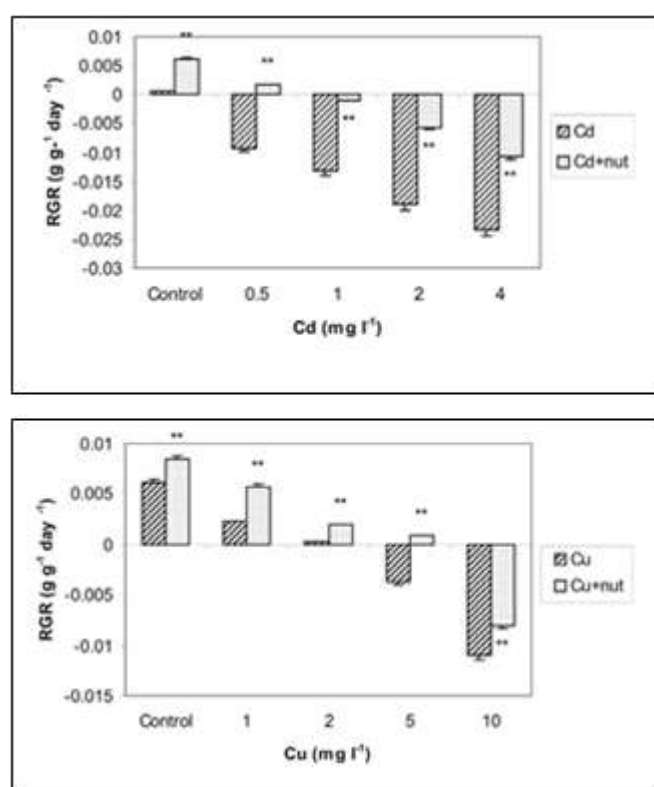
**Figure.1** Accumulation of (a) Cd (b) Cd  $\mu$  nutrients (c) Cu and (d) Cu  $\mu$  nutrients by *Gracilariaverrucosa* exposed to different concentrations over various periods of time. All values are means of triplicates + SD. ANOVA significance was set at  $P \leq 0.01$ . Different letters indicate significantly different values at a particular time point (DMRT,  $P \leq 0.05$ ).

The samples were centrifuged at 10,000 g for 10 min and absorbance was recorded at 532 and 600 nm. The amount of MDA was calculated by subtracting non-specific absorbance at 600 nm from absorbance at 532 nm. EC as a measure of ion leakage was determined according to Devi and Prasad (1998). Cadmium and copper exposed plants were washed with double deionized water and 500 mg of plant material was then transferred to 100 mL of deionized water for 24 h to facilitate maximum ion leakage. EC of the water was then recorded.

### III. Results

#### 3.1 Accumulation of heavy metals and effect on growth of plants

Plants accumulated high amounts of cadmium in a concentration-time dependent manner. The highest Cd accumulation was seen at a dose of 4 mg L<sup>-1</sup>. The percent of the total Cd accumulated (615 mgg<sup>-1</sup>) on day 7 at 4 mg L<sup>-1</sup> was 79.1% on day 1, 80.6% on day 3, and 92.8% on day 5 (Figure 1a). In the groups enriched with nutrients, Cd accumulation at 4 mg L<sup>-1</sup> was 87.9% on day 1, 90.8% on day 3, and 94.7% on day 5 of the total Cd accumulated (341.2 mgg<sup>-1</sup>) on day 7 (Figure 1b). Maximum Cu accumulation was observed at a dose of 10 mg L<sup>-1</sup>. The percent of the total Cu accumulated (351.6 mgg<sup>-1</sup>) on day 7 at 10 mg L<sup>-1</sup> was 72.7% on day 1, 81% on day 3, and 85.3% on day 5 (Figure 1c). In the groups enriched with nutrients, Cu accumulation at 10 mg L<sup>-1</sup> was 78.1% on day 1, 86.3% on day 3, and 90.9% on day 5 of the total Cu accumulated (110 mgg<sup>-1</sup>) on day 7 (Figure 1d). Relative growth rates of *Gracilariaverrucosa* decreased in the presence of cadmium in a concentration dependent manner (Figure 2). However, the treatments enriched with nutrients did not show a similar correlation (R=0.980). Relative growth rates of *Gracilariaverrucosa* decreased with copper in a concentration dependent manner (Figure 2), but the treatments enriched with nutrients did not show similar correlation (R=0.940). Therefore, nutrient addition attenuated the decline in relative growth rates in response to metal exposure (Figure 2).



**Figure 2** Relative growth rates of Cu and Cd treated plants (after 7 day). The bars represent standard deviation. \*\* (P < 0.01) indicate significant difference between the two treatment

### IV. Discussion

In the present study, high accumulations of Cd and Cu were observed in *Gracilariaverrucosa* over 7 days. Nutrient enrichment resulted in decreased accumulation of Cd and Cu in growing plants over the time. Growth rates of *Gracilariaverrucosa* decreased with increasing cadmium and copper concentrations. However, nutrient enrichment led to increased growth rates at Cd and Cu concentrations that impaired growth in the non-enriched groups. Cd had a stronger effect on growth rates than Cu. Low concentrations of Cu actually increased plant growth rates, while higher concentrations had negative effects.

### V. Conclusions

We have investigated the toxic effects of cadmium and copper on *Gracilariaverrucosa*, and showed that nutrient enrichment increased the tolerance of *Gracilariaverrucosa* to metal contamination. This effect has important implications in the use of constructed wetlands for industrial wastewater treatment. Many metallurgical industrial processes produce wastewater containing high concentrations of metal ions. Increased

tolerance may be useful for wastewater treatment by allowing macrophyte growth at metal concentrations that would otherwise impair their development. Nutrient addition will thus aid metal removal by increasing macrophyte production, leading to a higher metal uptake by the macrophyte biomass, and also by enhancing overall biological activity, reaching a higher metal retention in the detrital fractions.

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