1	Supplementary Information for
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3	Maltodextrin enhances biofilm elimination by electrochemical scaffold
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18	Running title: Maltodextrin and e-scaffold eliminates biofilms
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24 Electrochemical scaffold preparation. The purpose of the working electrode of the 25 electrochemical scaffold (e-scaffold) was to hold a negative polarity to reduce atmospheric oxygen and generate H₂O₂ (Sultana et al. 2015). To complete the electrochemical cell, we used a counter 26 27 electrode and a custom-made Ag/AgCl reference electrode. Briefly, a custom-built e-scaffold was 28 fabricated using carbon fabric (Panex 30 PW-06, Zoltex Companies Inc., St Louis, MO). The fabric was cut into a circular shape (6.42 cm²) to serve as the e-scaffold, and a smaller circular 29 30 carbon fabric "patch" (2.14 cm²) was used as the counter electrode. The counter electrode was 31 attached to the e-scaffold using a thin layer (~1 mm) of silicone rubber (DAP Dynaflex 32 230 Premium Indoor/Outdoor Sealant, catalog #18357), which provided insulation between the electrodes while still allowing oxygen to diffuse to the bottom surface of the e-scaffold for H₂O₂ 33 34 generation.

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36 For the controlled generation of H₂O₂, precise, accurate control of the potential of the e-scaffold 37 is essential (Istanbullu et al. 2012), and this was achieved using a Gamry Series G 300TM 38 potentiostat (Gamry Instruments, Warminster, PA, USA) in conjunction with a saturated Ag/AgCl 39 reference electrode (Lewandowski and Beyenal 2013). Ti wires (0.025 Ti, Malin Co., Cleveland, 40 OH, Lot #27567) were used as external connections to the potentiostat (Figure 1), and the 41 connection resistance was consistently $\leq 2 \Omega$. The e-scaffold was overlaid onto biofilms grown *in* 42 vitro on glass surface (Figure 1). This configuration allowed the ventral surface of the e-scaffold 43 to be exposed directly to biofilms.

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48 Fig. S 1. Growth curve of A) A. baumannii and B) S. aureus in presence of maltodextrin

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51 **References**

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