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exhibition at the Geological

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The chemistry of smartphones and other daily life items – the applied mineralogy exhibition at the Geological Survey of Norway

Geological Survey of Norway P.O. Box 6315 Torgarden NO-7491 Trondheim, Norway Tel. +47 73 90 40 00

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Summary: The report describes the contents and purposes of the applied mineralogy exhibition at the Geological Survey of Norway (NGU) in Trondheim. The exhibition comprises 446 objects currently exhibited in the common room of the Mineral Resources Division of the Geological Survey. The exhibition is structured according to economic commodity classes: metal ores and minerals, industrial minerals, and energy minerals. The metal ores and minerals are subdivided into precious metals (Au and Ag), base metals (Fe, Pb, Cu, Mn, Zn, Co, Ni), and rare metals (Sn, W, Sc, Li, Cs, and Be). The exhibited industrial minerals include quartz, feldspars, graphite, olivine, kyanite, micas, apatite, diamond, garnet, calcite, dolomite, kaolinite, montmorillonite, sulphur, and Ti minerals (Ti minerals, such as ilmenite and rutile, are considered as industrial minerals even Ti is a metal). The energy minerals include U-Th minerals. In addition, crude oil and heavy water are exhibited together with the energy minerals.

The purposes of the exhibition are, firstly, to display representative samples mainly from Norwegian but also from some international metal and industrial mineral deposits and, secondly, to illustrate how minerals are processed and in which products they are used. To demonstrate the value chain from mineral mining to final products, mineral concentrates and mineral products are displayed for some commodities. The target groups are visitors of the Geological Survey of Norway, such as politicians, researchers, school classes and public visitors.

In addition, the report provides chemical analysis of daily-life items, such as smartphones, compact cameras, toothpaste, deodorants and eye shadows to illustrate their complex chemistry and the diversity of minerals required for their production. The discussed examples of daily life items and the establishment of the applied mineralogy collection at NGU aim to raise awareness about the complexity of sourcing mineral raw materials and the need of mining.

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1. INTRODUCTION

In 2009 the idea arose to establish an exhibition of ore and industrial minerals and their industrial applications at the Geological Survey of Norway (NGU) in Trondheim. The exhibition has two major purposes, firstly, to display representative samples mainly from Norwegian but also from some international metal and industrial mineral deposits and, secondly, to illustrate how minerals are processed and in which products they are used. To demonstrate the value chain from mineral mining to final products, mineral concentrates and mineral products are displayed for some commodities. The target groups are visitors of the Geological Survey of Norway, such as politicians, researchers, school classes and public visitors. The exhibition is dynamic, meaning it undergoes continuous updating. By the end of 2015, the collection comprised 446 specimen and items (Appendix 1).

Besides documenting the collection content and purpose, this report includes chemical analysis of daily-life products, such as smartphones, compact cameras, toothpaste, deodorants, and eye shadows to illustrate their complex chemistry and diversity of minerals required for their production. The data provided is complementary to the exhibition.

2. STRUCTURE AND PURPOSE OF THE EXHIBITION

Most of the 446 objects of the collection are currently exhibited in glass cabinets in the meeting room of the Section for Mineral Resources at the Geological Survey of Norway. Some of the larger specimens (>1 kg) are kept in the storage room of the section. The exhibition is structured according to economic commodity classes: metal ores and minerals, industrial minerals, and energy minerals. The metal ores and minerals are subdivided into precious metals (Au and Ag), base metals (Fe, Pb, Cu, Mn, Zn, Co, Ni), and rare metals (Sn, W, Sc, Li, Cs, and Be). The exhibited industrial minerals include quartz, feldspars, graphite, olivine, kyanite, micas, apatite, diamond, garnet, calcite, dolomite, kaolinite, montmorillonite, sulphur, and Ti minerals (Ti minerals, such as ilmenite and rutile, are considered as industrial minerals even Ti is a metal). The energy minerals include U-Th minerals. In addition, crude oil and heavy water are exhibited together with the energy minerals.

The focus is on Norwegian deposits, but the exhibition does not aim to show examples of every Norwegian deposit. For commodities which have not been mined in Norway, examples from other countries are shown. Most of the rock and mineral specimens have fist size and are relative compact and not so easily breakable, so that visitors and users of the exhibition are allowed to take specimen in their hands to inspect them. For several ores and minerals, processed mineral concentrates and daily life products are exhibited. Examples of the exhibited objects are shown in Figures 1 to 4. The collection and exhibition are dynamic, meaning that specimens can be added, and the exhibition themes changed.

In the following, chemical analyses of some daily life items are provided to illustrate which minerals and in which amounts are needed for their production. This aims to demonstrate the use and need of minerals in industrial production, and to raise societal awareness and acceptance of mineral exploration and mining.

Figure 1. Exhibition display showing the industrial minerals olivine and kyanite, their processed concentrates (in glasses) and one application (olivine refractory brick used for smelter paneling). Source: NGU.

Figure 2. Exhibition display showing the base metal minerals cassiterite (SnO2), three examples of Sn concentrates (glasses to the left) and one application (tin cup). Source: NGU.

Figure 3. Exhibition display showing the industrial minerals calcite (limestone, marble) and kaolinite (glasses), one processing product (Hydrocarb paper filler) and one application (paper). Source: NGU.

Figure 4. Exhibition display showing Fe minerals and ores, processed iron pellets from the Kiruna mine in Sweden (small glass) and two Fe products comprising Fe-oxide pigment "Falun Red" (in glass) and a large steel bolt. Source: NGU.

3. METHODS APPLIED FOR THE CHEMICAL ANALYSIS OF DAILY-LIFE ITEMS

3.1 Comminution of compact cameras and smartphone

Three compact digital cameras (models: Fujifilm Finepix JZ, Canon IXUS 400, and Nikon Coolpix 7900) and one smartphone (LG Optimus L7 on the market since 2012) were analysed for bulk composition to reveal their chemical complexity and the diversity of minerals needed to manufacture these devices (Figures 5 and 6). Chemical analysis requires that the devices are crushed and milled ideally to a particle size <200 µm.

However, comminution of individual, small electronic devices into a homogenous, equigranular powder for high precision chemical analysis is still a technical challenge. In a previous study, Müller (2013) comminuted mobile phones by electrical fragmentation (Selfrag, 2015). The highvoltage, pulsepower fragmentation of mobile phones, which was performed in a liquid, resulted in a wide range of particle sizes ranging from powder $(-20 \mu m)$ to nearly undestroyed metal and plastic parts up to 10 cm in size. Multiple composed components of the mobile phones such as printed circuit boards and micro-processors were well fragmented and partially powdered. Large single-material components, such as the plastic body and supporting metal frame were hardly damaged. This is because electronic fragmentation weakens the object primarily along particle and component boundaries. Thus, single-material components without internal boundaries are hardly affected by electronic fragmentation. Components >1 cm were therefore cut manually down to <1 cm. Because electronic fragmentation was performed in a liquid, the powdered material was suspended and had to be filtered. The process resulted in a weight loss of 7-10 wt.% of the material adding to the uncertainties of the chemical results.

Figure 5. Compact cameras which were crushed and analysed in the frame of this study. Source: NGU.

Figure 6. Smartphone (type LG Optimus L7) which was crushed and analysed in the frame of this study. Source: NGU.

Figure 7. Compact cameras set in liquid nitrogen for 30 minutes prior to crushing. Source: NGU.

In this study, comminution of compact cameras and the smartphone was carried using conventional, laboratory-scale crushes and shredders at the Mineral Processing Laboratory at NTNU in Trondheim, Norway. The workflow included several steps. First, prior to crushing, compact cameras and the smartphone were set into liquid nitrogen for about 30 minutes to increase their fragmentarily [\(Figure 7\)](#page-11-1). In the next step, the electronic devises were crushed with an 8A Hazemag Impact Crusher type D-4400 [\(Figure 8A](#page-13-0)). Results of the crushed devises are shown in Figures 9A to 9D. The final steps involved cutting of the crushed material by a Retsch SM 2000 shredder using bottom sieves with 8 mm mesh size [\(Figure 9E](#page-14-0)) and 2.5 mm mesh size [\(Figure 9F](#page-14-0)). Micro magnets from speaker, receivers, vibration mode motor, tactile feedback motor, and camera autofocus mechanism of the smartphone sticked to the inside walls of the used machines and were collected with a strong magnet. Because of the complex comminution, loss of material could not be avoided. The weight of lost material was considered when evaluating the precision of element concentrations determined by the following chemical analysis [\(Table 1\)](#page-12-0).

Table 1. Original weight and weight loss during crushing and shredding of electronic devices.

Figure 8. Crushers used for crushing compact cameras and mobile phones. A - Hazemag Impact Crusher for dry material 0-100 kg/h. B - Retsch SM 2000 Shredder. Source: NGU.

Figure 9. Smartphone (A) and compact cameras (B, C, D) after crushing with the Hazemag Impact crusher. The crushed compact camera shown in D after grinding with the Retsch SM 2000 shredder using first a bottom sieve with 8 mm mesh size (E) and afterwards a bottom sieve with 2.5 mm mesh size (F). Source: NGU.

3.2 Chemical analysis of compact cameras, smartphone, eye shadows, and deodorants

The chemical analyses of compact cameras, smartphone, eye shadows, and deodorants were performed at Activation Laboratories Ltd. in Ancaster, Ontario, Canada (Actlabs, 2015). A combination of several methods was applied to get an almost complete list of analysed chemical elements. The methods included instrumental neutron activation analysis (INAA) for the elements Ag, Br, Cl, Ir, Lu, Sc, Yb, Zn, and Cr, infrared spectroscopy for C and S, inductively coupled plasma optical emission spectroscopy (ICP-OES) and inductively coupled plasma mass spectrometry (ICP-MS) for the remaining major and trace elements. In the case of deodorants, element concentrations were determined by solution ICP-MS and ion chromatography (Cl, Br, F, N, S, P). The used instrumentations and methodologies are summarized in [Table 2.](#page-15-2) The relative analytical error of the determined element concentration is 5%. Elements not included in the analytical package are Na, F, N, H, noble gases, Po, At, Fr, Ra, Ru, Rh, Tc, Re, Os, Ac, Pa, and O.

3.3 Ion chromatography for the analysis of F in toothpaste

Fluorine (F- ion) content in toothpaste was determined using the chromatographic system Dionex ICS-1100 connected to a Dionex AS-DV auto sampler and a Chromeleon 7.1 workstation at the Geological Survey of Norway (NGU) in Trondheim. Details about the analytical system used are provided in [Table 3.](#page-16-2) The standard solutions listed in [Table 4](#page-16-3) were diluted with water type 1 (18.2 MΩ.cm) to the appropriate level to calibrate the system to measure fluoride, chloride, nitrite,

bromide, nitrate, phosphate, and sulphate. Approximately 0.7 g toothpaste was dissolved in 45 ml water type 1 (18.2 MΩ.cm), and an ultrasonic bath was used to complete the dissolution. 1 ml of this suspension was further diluted up to 5 ml with water type 1 in a 5 ml sample tube with 20 μ m filter cap prior analysis, with a total dilution factor between 2800 and 3200.

Unit	Description
Column	Ion Pac AS 14 A 4x250 mm P/N 56904
Guard column	Ion Pac AG 14 A 4x250 mm P/N 56897
Suppressor	ASRS 300 4-mm P/N 064554
Injection loop	25 µl
Temperature	30° C
Ovn	ICS-1100 Column Heater
Eluent	0.8 M Na ₂ CO ₃ + 0.1 M NaHCO ₃
Sample tube	5 ml Poly Vials with 20 um filter caps P/N 038141

Table 3. Instrument conditions of the chromatographic system Dionex ICS-1100.

4. Results

4.1 Chemistry of compact digital cameras and smartphones

Compact digital cameras are small and lightweight, making them easy to carry around for everyday photography. Even if the use of compact digital cameras has strongly declined because of the increasing quality of in-built cameras in smartphones, compact cameras are still preferred by many costumers due to the higher image quality (because of larger sensors and more advanced optics), optical zoom lenses that can physically zoom in on subjects (smartphones typically rely on digital zoom, which compromises image quality), more manual control options allowing photographers to have greater creative control over their images, and better device handling and ergonomics. The construction of compact digital cameras includes a metal frame, built-in lens, built-in flash, sensor that captures light and converts it into digital information, processor for image processing, built-in storage, electronics to connect to external devices, battery and viewing screen for framing and reviewing photos. Each of these parts consists of many components consisting of different materials [\(Figure 10\)](#page-17-0).

Figure 10. Example of a disassembled compact digital camera in this case a Canon PowerShot A530 camera. Source: Wikimedia.

Today smartphones are some of the most used electronic devices. The in-built camera, flash, processor etc. in smartphones have much smaller dimensions than in compact cameras. In addition, smartphones have a large liquid crystal display (LCD), circuit board, antenna, microphone, speaker and a relatively large battery. These parts are made of many individual components of complex chemical composition. In all, they contain 75 % metals and 25 % plastics. However, looking in detail, the chemistry of smartphones is rather complex. For the casing, alloys of Fe, Al, Mg and Ni are used. Nickel is used as well in the microphone diaphragm, electrical connections, capacitors and battery. Beside Ni, C, Co, Li and Br are constituents of the battery. Ag, Au, and Cu are the major metals in microelectronic components. Tantalum is the major component of micro capacitors. Indium is part of the LCD display where it is used as Sn-doped In_2O_3 . Iron alloys containing Nd, Pr and Gd are used in the magnets contained in the speaker, microphone, vibrator and micro motors. [Figure 11](#page-18-0) illustrates the complex chemical composition of just one component: the vibrator. The vibrator alone is made up of parts made of 10 different metals: Fe, Ni, Cu, Cr, Au, Ag, W, Pd, Nd and Pr. However, complete and precise chemical bulk analyses of individual smartphones are not readily available. Several sources provide information on the major and some trace element compositions of smartphones. One of the few comprehensive studies which provided an almost complete list of elements in mobile phones was done by Müller et al. (2013).

In the frame of this study, three compact cameras including a Fujifilm Finepix JZ, Canon IXUS 400 and a Nikon Coolpix 7900 were grinded and analysed to reveal their chemical complexity. The results of the chemical analysis are shown in [Table 5.](#page-19-0) The relative analytical errors of the determined concentrations are ±9% for Canon, ±13% for Nikon, ±11% for Fujifilm, and ±15% for LG. About 49 elements out of 68 determined elements occur with concentration >1 mg/kg in the analysed devices.

In the diagrams in Figures 12 and 13, elements are ordered according to their abundance in the Canon camera. The most abundant, major (>0.4 wt.%/ >4000 mg/kg) elements in these devices are (from higher to lower contents) C (24-30 wt.%), Cu (7-16 wt.%), Fe (12-16 wt.%), Al (3.7-11.8 wt.%), Ni (0.4 to 6.1 wt.%), Cr (3.4-6.2 wt.%), Si (2.3-5.5 wt.%), Ca (0.4-1.6 wt.%), Zn (0.29-0.77 wt.%), Mn (0.05-0.52 wt.%), and Sn (0.18-0.56 wt.%). Carbon as the most abundant element is bound in polymers of the plastic parts and used as elementary carbon in the batteries. Overall, the compositions of the three cameras and the smartphone are relatively similar. There are, however, device-specific differences (see element concentrations marked in red in [Table 5\)](#page-19-0). The Canon camera is significantly richer in Mo (3150 mg/kg), Ni (6.1 wt.%), Pb (1850 mg/kg), S (1590 mg/kg), Sc (10 mg/kg), Ta (355 mg/kg), V (420 mg/kg) and W (412 mg/kg) and contains lower Y (4.5 mg/kg) and Mg $\left($ <100 mg/kg). The Nikon camera has high Al (11.8 wt.%) and Ba (5410 mg/kg) and the Fujifilm camera high Bi (107 mg/kg), Sb (798 mg/kg), Nd (1590 mg/kg) and Pr (424 mg/kg), and low in Mn (502 mg/kg). The smartphone is higher in Ni (5.4 wt.%) and has low Br (17 mg/kg), La (97 mg/kg) and Gd (12 mg/kg). The low Br in the smartphone is due to the different battery type used in the phone.

When compared to the average composition of the upper continental crust (UCC; Figures 14 to 16), the part of the Earth from where minerals are sourced, Au (51-153 mg/kg) is about 100,000 times enriched in the devices. The average concentration of Au in the UCC is 0.0015 mg/kg. Silver (259-428 mg/kg), Cu (8.5-16.2 wt.%) and Pd (2-5 mg/kg) are about 10,000 times enriched. Other elements which are considerably enriched and not named above include W (18-412 mg/kg), Mo (24-3150 mg/kg), Sb (21-798 mg/kg), Ta (60-355 mg/kg), Se (10-495 mg/kg), In (4-21 mg/kg) and the REEs La (97-1480 mg/kg), Gd (12-262 mg/kg), Pr (3-424 mg/kg) and Dy (3-33 mg/kg).

[Table 5](#page-19-0) also lists the minerals from which the different elements are sourced through mining and mineral processing. The major producers of these minerals in 2015 are listed in [Table 5](#page-19-0) as well. Even though most minerals are mined in China, there is a great diversity of countries from which the individual ore minerals origin. The list of countries illustrates the complexity of the product supply chains considering only the raw material sources for compact cameras and smartphones. This is one reason why the device producers have little or no control or knowledge about the raw material sources.

Figure 11. Example of a construction of a vibrator found in smartphones. The motor is 10 mm in diameter. The lower-case numbers next to the element abbreviations are concentrations in wt.% with an analytical error of 5 wt.%. Ten different metals, including Fe, Ni, Cu, Cr, Au, Ag, W, Pd, Nd and Pr are beside plastic required to manufacture the vibrator. Modified from Müller (2014).

Table 5. Results of chemical analysis of compact cameras (Canon, Nikon, Fujifilm) and smartphone (LG). Element concentrations are provided in mg/kg. Relative analytical errors of the determined concentrations are ±9% for Canon, ±13% for Nikon, ±11% for Fujifilm, and ±15% for LG. Numbers in red mark concentrations which very are different to the other analysed devices. Major ore minerals are provided for elements only, which have considerable concentrations in the analysed which very are different to the other analysed devices. Majo **devices. UCC – Average concentration of the element in the upper continental crust according to Rudnick and Gao (2014). n.p. – not provided; DL – Detection limit.**

Figure 12. Column chart of the 25 most abundant elements in compact cameras and smartphone. Elements are ordered according to their abundance in the Canon camera.

Figure 13. Column chart of less abundant elements in compact cameras and smartphone. Elements are ordered according to their abundance in the Canon camera.

Figure 14. Element concentrations of smartphones normalised to the average composition of the Earth's upper crust according to Rudnick and Gao (2014). The diagram illustrates how elements in smartphones are enriched (>1) by natural processes (in the mineral deposit which was mined) and by industrial processing (rock crushing, mineral separation, metallurgy) or depleted (<1) compared to the UCC.

Figure 15. Column chart of less abundant elements in compact cameras and smartphone. Elements are ordered according to their abundance in the Canon camera.

Figure 16. Column chart of less abundant elements in compact cameras and smartphone. Elements are ordered according to their abundance in the Canon camera.

4.2 The chemistry of decorative cosmetics: Example eye shadow

Eye shadow is a decorative cosmetic applied to the eyelids to attract attention. The manufacturing and use of eye shadows dates back to predynastic Egypt about 5000-4000 BCE (Lucas, 1930; Kreston, 2012). Today, eye shadow or eye makeup is chemically and mineralogical rather complex. Eye shadows typically consist of five ingredient types: base fillers, binders, slip, color pigments and preservatives. Base fillers are usually minerals such as mica, talc or kaolin, which add bulk and texture to eye shadow. Mica absorbs moisture, gives the eye shadow shine and luster, and makes it opaque. Binders help eye shadow adhere and stay attached to skin. Zinc and magnesium, which are both white powders, are commonly used as dry binders. Silicone, paraffin wax, mineral oil or vegetable oils may be used as liquid binders. Slip allows eye shadow to glide across the skin smoothly. Products may use silica or nylon, which are fine, colorless powders. Other types of slip include dimethicone, boron nitride or bismuth oxychloride. The list of color pigments used can be long depending on the color to be achieved. Preservatives help products stay bacteria free and extend their lifespan. Common preservatives in eye shadow are glycol and tocopherol.

In [Table 6](#page-26-0) an example of list of ingredients in the eye shadow "01 Scooby blue - 4 trendige Farben" produced by the German brand Rival de Loop Young is provided [\(Figure 17\)](#page-25-0). The 28 ingredients are made of 21 different chemical elements. Eighteen of the elements are sourced from 18 different minerals, which are shown in [Table 6](#page-26-0) together with the major mining country of these minerals. Because the actual mining country of a particular mineral is difficult to trace, the major producing country is provided in [Table 6](#page-26-0) to show the complexity of the raw material supply. Carbon (±H) is sourced from crude oil or plant- or animal-based raw materials, and N and O from air. Bromine is extracted from Br-rich brines.

Half of the ingredients are color pigments. The color pigments used for this product, as provided by the producer, are bismuth oxychloride (BiOCl), tin oxide $(SnO₂)$, titanium oxide (TiO₂), ultramarine (lapis lazuli rock), chromium oxide greens (Cr₂O3), ferric ferrocyanide (C₁₈Fe₇N₁₈), two variants of iron oxide (Fe₂O₃), manganese violet (NH₄MnP₂O₇), tartrazine (C₁₆H₉N₄Na₃O₉S₂), lithol rubin (C₁₈H₁₂CaN₂O₆S), acid red (C₂₀H₂Br₄Cl₄Na₂O₅), and allura red (C₁₈H₁₄N₂Na₂O₈S₂) [\(Table 6\)](#page-26-0). For the production of these pigments alone, 13 different minerals are needed. Additionally, mica and kaolin may be considered color pigments, because mica causes the glitter effect and kaolin improves the opaqueness.

[Table 7](#page-27-0) lists element concentrations analysed in four eye shadows ("White", "Bright blue", "Steel blue bright", "Steel blue dark") with different color from the Rival de Loop Young company (see also [Figure 17\)](#page-25-0). In the diagrams in Figures 18 and 19, elements are ordered according to their abundance in the eye shadow "White". In general, the determined element concentrations are relative similar for all four eye shadows. The most abundant elements (beside O and H) are Si (16.9-20.2 wt.%), C (8.8-15.3 wt.%), Ti (4.6-13.0 wt.%), Mg (7.4-9.6 wt.%), Al (4.4-5.8 wt.%), K (1.8-2.4 wt.%), Fe (0.7-1.8 wt.%), Ca (786-2666 mg/kg), F (2300-2800 mg/kg), and Bi (1->2000 mg/kg). The Bi content in the eye shadow "White" is higher than the upper detection limit of the applied ICP-MS methodology. However, the other three eye shadows have much lower Bi contents between 1 and 9 mg/kg. Other major differences in chemistries among the investigated eye shadows are high Cr (105 mg/kg) and low Ti (46282 mg/kg) in "Steel blue bright", high Sn (1544 mg/kg) in "Steel blue dark", low Cr (<5 mg/kg) in "White", and low V (<8 mg/kg) in "Bright blue". These differences are caused by differences in the use color pigments and their mixtures. Other elements with concentrations between 2000 and 10 mg/kg are Na, Cl, Rb, P, Ba, Sn, Mn, Nb, W, Cr, Ga, Cs, V, and Zr [\(Table 7\)](#page-27-0).

Barium, Cs, F, Ga, Nb, Rb, W, V and Zr are elements which were determined with considerable concentrations (>10 mg/kg) but are not listed in the ingredients list provided by the producer (compare Tables 6 and 7). The reasons for that are mainly the used minerals, because natural minerals incorporate not only the major elements they are made of, but also traces of other elements which are considered natural impurities or contaminations. Cesium (16-28 mg/kg), F (2300-2800 mg/kg) and Rb (197-354 mg/kg) are most likely trace elements in the used mica. Gallium (16-29 mg/kg) is a trace constituent of kaolin, W (20-33 mg/kg) and Nb (23-57 mg/kg) of cassiterite, V (<8-23 mg/kg) of chromite, and Zr (5-22 mg/kg) of rutile.

Figure 17. Eye shadow powders of the product "Colour Palette" of German brand Rival de Loop Young. For analysis, the four eye shadows of "01 Scooby blue" palette shown in the upper left were chosen. Source: NGU.

Table 6. Ingredients in the eye shadow "Scooby blue - 4 trendige Farben" produced by the German brand Rival de Loop Young. Minerals, which are needed to produce the ingredients, are listed and the major mining country as well. Mineral-based color pigments are shown in red.

Table 7. Analytical results of four eye shadows of "Scooby blue - 4 trendige Farben" produced by the German brand Rival de Loop Young. Numbers in red mark concentrations which very are different to the other analysed eye shadows.

Figure 18. Column chart of the 16 most abundant elements in eye shadows. Elements are ordered according to their abundance in eye shadow "white", which is richer in Bi than the other analysed eye shadows.

Figure 19. Column chart of the 16 most abundant elements in eye shadows. Elements are ordered according to their abundance in eye shadow "white".

4.3 The chemistry of deodorants

Deodorants are a mixture of antiperspirants, antibacterial ingredients, fixatives, moisturizer, emollients, emulsifier, stabilisers, humectants, propellants, additives for pH control, perfumes and preservatives (e.g., Shen and Nardello-Rataj, 2009). The functional most important ingredients are antiperspirants, which inhibit perspiring and sweating, and antibacterial ingredients which kill bacteria and neutralise smell. Antiperspirants are commonly aluminium compounds such as aluminium chloride, aluminium chlorohydrate, aluminium hydroxybromid, and aluminium zirconium trichlorohydrate glycine. The aluminium ions released from these compounds form a temporary blockage in the sweat ducts, stopping the flow of sweat to the skin's surface (e.g., Laden, 1999). Common antibacterial ingredients are alcohol, triclosan, ethylhexylglycerin, polyglyceryl-2 caprate, chlorhexidine, polyhexamethylene, aluminium salts, zinc salts or zinc oxide. Deodorants are available as spray, roll-on or sticks. The concentration of ingredients

increases from spray, to roll-on and stick. [Table 8](#page-31-0) provides an example of ingredients of the Sterilan 'Men Extra Cool' roll-on deodorant are listed as provided by the producer. The 25 ingredients are composed of ten different chemical elements. Five of the ten elements, Al, Si, Cl, Zr and Na, are sourced from four different minerals. In addition, the minerals quartz and talc are directly used in powdered form.

In the frame of this study, two deodorants sprays, Dior 'Homme Sport' and Old Spice 'Pure Sport', and two roll-on deodorants, Sterilan 'Men Extra Cool' and Palmolive 'Men Pure Arctic', have been analysed [\(Figure 20\)](#page-30-0). The aim was to determine the absolute concentrations of the elements in these deodorants. The results of the chemical analyses are given in [Table 9.](#page-32-0) Elements which were detected in the deodorants including (from high to low concentrations): Cl, Al, Zr. Hf, S, N, F, Zn, Ti, Na, Fe, Cd, Sc, Ga, V, and Te [\(Figure 21\)](#page-30-1). Both roll-ons, Sterilan and Palmolive, have the highest concentrations of most of these elements, except F which is highest in Dior, and Zn which is highest in Old Spice.

Sterilan and Palmolive are particularly enriched in Cl (19000 and 31400 mg/l respectively), Al (about 4000 mg/l), but very different in the Zr content (1660 and 0.03 mg/l, respectively). The difference in Zr is because Sterilan contains aluminium zirconium tetrachlorohydrex glycine and Palmolive aluminium hydrochlorohydrate as antiperspirant. Because the mineral zircon is the source of Zr in the Sterilan antiperspirant, relatively high Hf concentrations of 40.7 mg/l were detected. Hafnium is a common trace element in natural zircon and, thus, has been inherited in the antiperspirant from the original raw material. The same inheritance process applies for Ga found Sterilan and Palmolive. The elevated Ga content of about 0.3 ml/g in the roll-ons is due to the use of bauxite as raw material for the production of the antiperspirant. Clay minerals in bauxite are naturally enriched in Ga compared to other minerals and, thus, Ga in the antiperspirant is inherited (without purpose) from the raw material. The high S content in Sterilan is unexpected, because according to the ingredient list provided by the producer, there is no Sbearing ingredient. Also, there is no straightforward explanation for the elevated concentrations of Cd, Sc, V, and Te in Sterilan. One explanation could be, that chemical catalysts containing these elements, were used for ingredient production. The Old Spice deodorant is particularly enriched in Zn (8.6 mg/l). This is because of the use of zinc phenolsulfonate $(Zn(C_6H_5O_4S)_2$ as antibacterial ingredient. Dior has the lowest concentrations of all elements compared to the other investigated deodorants, except for F (12.5 mg/l). Dior is the only analysed deodorant which contains traces of F. However, none of the ingredients on the list provided by the producer, contains F.

Figure 20. Deodorants analysed in the frame of this study. Source: NGU.

Table 8. Ingredients in the deodorant Sterilan 'Men Extra Cool' as provided by the producer. Minerals, which are needed to produce the ingredients, are listed and the major mining country as well. Mineralsourced elements and mineral ingredients are shown in red.

011 centrations are given in my/i. Analyte Symbol	Unit Symbol	Detection Limit	Analysis Method	Sterlian Roll-on	Palmolive Roll-on	Old Spice Spray	Dior Spray
Ag	mg/l	0.04	ICP-MS	< 0.04	<0.04	<0.04	<0.04
Al	mg/l	0.4	ICP-MS	4030	3820	< 0.4	< 0.4
As	mg/l	0.006	ICP-MS	< 0.006	< 0.006	< 0.006	0.006
Ba	mg/l	0.02	ICP-MS	<0.02	< 0.02	< 0.02	<0.02
Be	mg/l	0.02	ICP-MS	<0.02	<0.02	<0.02	<0.02
Bi	mg/l	0.06	ICP-MS	<0.06	<0.06	<0.06	<0.06
Br	mg/l	3	IC	3	$<$ 3	-3	$<$ 3
Сa	mg/l	100	ICP-MS	< 100	< 100	< 100	100
Cd	mg/l	0.002	ICP-MS	2.04	< 0.002	< 0.002	<0.002
Сe	mg/l	0.0002	ICP-MS	0.0020	0.0022	< 0.0002	< 0.0002
CI	mg/l	0.03	IС	31400	19000	14	10
Co	mg/l	0.001	ICP-MS	0.00114	< 0.001	< 0.001	< 0.001
Cr	mg/l	0.1	ICP-MS	< 0.1	< 0.1	0.1	< 0.1
Сs	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
Cu	mg/l	0.04	ICP-MS	< 0.04	<0.04	<0.04	<0.04
Dy	mg/l	0.0002	ICP-MS	0.0004	< 0.0002	<0.0002	< 0.0002
Er	mg/l	0.0002	ICP-MS	0.0006	< 0.0002	< 0.0002	< 0.0002
Eu F	mg/l	0.0002 1	ICP-MS IC	< 0.0002 <1	< 0.0002	< 0.0002	< 0.0002
Fe	mg/l mg/l	$\overline{2}$	ICP-MS	2.05	<1 $<$ 2	<1 <2	12.5 $<$ 2
Ga	mg/l	0.002	ICP-MS	0.292	0.342	< 0.002	<0.002
Gd	mg/l	0.0002	ICP-MS	0.0002	< 0.0002	< 0.0002	< 0.0002
Ge	mg/l	0.002	ICP-MS	< 0.002	< 0.002	< 0.002	< 0.002
Hf	mg/l	0.0002	ICP-MS	40.7	0.00168	< 0.0002	< 0.0002
Hg	mg/l	0.04	ICP-MS	<0.04	< 0.04	< 0.04	< 0.04
Ho	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
In	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
Κ	mg/l	6	ICP-MS	-6	<6	<6	<6
La	mg/l	0.0002	ICP-MS	0.0036	0.0036	< 0.0002	0.0002
Li	mg/l	0.2	ICP-MS	<0.2	<0.2	<0.2	< 0.2
Lu	mg/l	0.0002	ICP-MS	0.0002	< 0.0002	< 0.0002	< 0.0002
Mg	mg/l	0.4	ICP-MS	< 0.4	< 0.4	< 0.4	< 0.4
Mn	mg/l	0.02	ICP-MS	0.030	0.045	<0.02	<0.02
Mo	mg/l	0.02	ICP-MS	<0.02	<0.02	<0.02	< 0.02
Na	mg/l	1	ICP-MS	2.97	56.4	<1	<1
Nb	mg/l	0.001	ICP-MS	0.0132	< 0.001	< 0.001	< 0.001
Nd Ni	mg/l mg/l	0.0002 0.06	ICP-MS ICP-MS	0.0006 <0.06	0.0004 <0.06	< 0.0002 <0.06	< 0.0002 0.06
N	mg/l	0.01	IС	16.1	6.7	12.2	4.4
Pb	mg/l	0.01	ICP-MS	0.008	0.004	0.002	0.002
Pr	mg/l	0.001	ICP-MS	0.0002	< 0.0002	< 0.0002	< 0.0002
P	mg/l	$\overline{2}$	IC	<2	<2	<2	<2
Rb	mg/l	0.001	ICP-MS	< 0.001	< 0.001	< 0.001	< 0.001
Sb	mg/l	0.002	ICP-MS	< 0.002	< 0.002	< 0.002	< 0.002
Sc	mg/l	0.2	ICP-MS	0.5	<0.2	<0.2	<0.2
Se	mg/l	0.04	ICP-MS	< 0.04	< 0.04	< 0.04	< 0.04
Si	mg/l	40	ICP-MS	<40	<40	<40	<40
Sn	mg/l	0.02	ICP-MS	0.02	<0.02	<0.02	< 0.02
Sm	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
S	mg/l	3	IС	26.1	$<$ 3	<3	-3
Sr	mg/l	0.008	ICP-MS	< 0.008	< 0.008	< 0.008	< 0.008
Та	mg/l	0.0002	ICP-MS	0.002	< 0.0002	< 0.0002	< 0.0002
Tb	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
Тe	mg/l	0.02	ICP-MS	0.182	< 0.02	< 0.02	<0.02
Th Τi	mg/l mg/l	0.002 0.02	ICP-MS ICP-MS	0.0070 3.66	< 0.0002 0.09	< 0.0002 <0.02	< 0.0002 <0.02
TI	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
Tm	mg/l	0.0002	ICP-MS	< 0.0002	< 0.0002	< 0.0002	< 0.0002
U	mg/l	0.0002	ICP-MS	0.0026	0.0014	< 0.0002	< 0.0002
\vee	mg/l	0.02	ICP-MS	0.26	0.18	< 0.02	<0.02
W	mg/l	0.004	ICP-MS	< 0.004	< 0.004	< 0.004	< 0.004
Υ	mg/l	0.0006	ICP-MS	0.0129	< 0.0006	< 0.0006	< 0.0006
Zn	mg/l	0.1	ICP-MS	< 0.1	0.126	8.55	< 0.1
Zr	mg/l	0.002	ICP-MS	1660	0.032	< 0.002	<0.002
Yb	mg/l	0.0002	ICP-MS	0.001	< 0.0002	< 0.0002	< 0.0002

Table 9. Chemical analysis of deodorants. Elements with elevated concentrations are highlighted in red. Concentrations are given in mg/l.

4.4 Minerals in toothpaste

Toothpaste is a paste or gel of complex composition used with a toothbrush to clean and maintain teeth. Beside foaming agents and odorants, toothpaste contains a number of mineralbased ingredients for abrasion, preventing tooth decay and to colour the paste. [Table 10](#page-34-1) shows the ingredient list of the toothpaste Colgate 'Karies Kontroll' as provided by the producer Colgate-Palmolive Norge AS. This toothpaste has 18 ingredients consisting of 13 different chemical elements. These 13 elements represent 14% of the element diversity of our universe. Nine of these 13 elements are sourced from nine different minerals, which are mined in different parts of the world. In addition, the minerals quartz and mica are directly used in powdered form.

Fluorine-bearing substances in toothpaste are mineralogically the most interesting. The use of fluoridated toothpastes was first introduced in the 1970s. Today, over 90% of the toothpastes produced worldwide contain F-bearing substances. Fluorine in these substances is sourced from the mineral fluorite (CaF2). Fluorine substances promote the formation of dental enamel to prevent or reducing the incidence of dental caries and dental erosion. The strengthening of dental enamel by F is based on a simple chemical reaction. Dental enamel consists of the mineral hydroxylapatite ($Ca₅(PO₄)₃(OH)$). When reactive F- ions come into contact with the enamel tooth surface, OH- of hydroxylapatite is partially replaced by F- ions forming fluorapatite (Ca₅(PO₄)₃(F)). Because fluorapatite is more resistant to acid than hydroxylapatite, the modified enamel provides better protection against caries and other forms of damage. Fluorine is added to toothpaste typically at levels of 1000 to 1500 mg/kg. Concentrations below 1000 mg/kg are not likely to be preventive, and the preventive effect increases with concentration (Walsh et al., 2010).

In the frame of this study, three different toothpastes, Colgate 'Karies Kontroll', Solidox 'Total beskyttelse', and Elmex 'Kariesschutz', were analysed for F using ion chromatography. In addition to F, nitrate, phosphate or sulphate were detected [\(Table 11\)](#page-35-1). Different F-bearing substances are used in these pastes. Elmex contains amine fluoride, Solidox sodium fluoride, and Colgate sodium fluoride and sodium monofluorophosphate [\(Table 11\)](#page-35-1). For the Elmex toothpaste, the determined F of 1302 mg/kg is a bit lower than the content provided by the producer (1400 mg/kg). The Solidox toothpaste contains the highest F content with 1446 mg/kg. The F content detected by ion chromatography in the Colgate toothpaste of 402 mg/kg is much lower than the concentration given by the producer. That is because only F bound in the ingredient sodium fluoride was detected [\(Table 11\)](#page-35-1). The applied method was not able to detect F in sodium monofluorophosphate. The Cl, N, and S contents of the investigated toothpastes are highly variable. Solidox and Elmex are high in Cl 1342 and 1029 mg/kg, respectively) whereas Colgate has low Cl of 154 mg/kg. Also, S and N is low in the Colgate paste compared to the Solidox and Elmex pastes. Solidox contains sodium lauryl sulfate ($NaC₁₂H₂₅SO₄$) and sodium saccharin (C₇H₅NNaO₃S⁺) which explains the high S and N.

Figure 22. Sketch illustrating the dissolution of dental enamel (hydroxylapatite) by acids, and the substitution of OH-ions by F-ions released from the toothpaste to form fluorapatite at the enamel's surface. Source: NGU.

Table 11. Fluorine contents in three toothpastes as provided by the producer and analysed with ion chromatography in the frame of this study. n.a. – not analysed, n.p. – not provided by the producer. Concentrations are given as mg/kg.

Toothpaste type	Colgate 'Karies Kontroll'	Solidox 'Total beskyttelse'	Elmex 'Kariesschutz'
F-bearing ingredient as provided by producer	Sodium monofluorophosphate (1000 mg/kg F) and sodium fluoride (450 mg/kg F)	Sodium fluoride	Amine fluoride
F content as provided by producer	1450	n.p.	1400
F content determined by ion chromatography	402	1446	1302
CI.	154	1342	1029
N bound as nitrate	691	30956	1196
P	38878	n.a.	n.a.
S	201	3227	633

5. CONCLUSIONS AND OUTLOOK

Analyses of daily life items performed in the frame of this study, including compact cameras, smartphone, eye shadows, deodorants, and toothpastes, illustrate their complex chemistry. The production of components and ingredients of these consumer goods requires a wide range of mineral-based raw materials (i.e. minerals), as well as plant-, animal-, and crude-oil-based raw materials. The minerals have to be mined in different countries around the world. [Table 12](#page-35-2) summarizes how many minerals are needed for the production of the items mentioned above and the number of mining countries. Small amounts of certain mineral-based raw materials, such as Fe, Au, Ag, Al, Pb, and Cu, may have been sourced from recycled waste. Mineral processing may occur in the same country where the mineral is mined, but this is not always the case. The next step in the production cycle, the manufacture of chemical ingredients and components, commonly takes place in different countries. Consequently, the production cycle from mining through mineral processing, manufacture of ingredients and components to final product assembly, and transport, logistics and trading is very complex.

The main aim of this study and the establishment of the collection at NGU is to raise awareness about the complexity of sourcing mineral raw materials, the need for mining and the need to address associated environmental and social impacts. The increasing societal demand for better living conditions, combined with a steadily growing global population, drives the need for higher production. Mining of minerals will remain essential, even as waste recycling becomes more prevalent. However, recycling of deodorants and toothpaste, for example, is not technological feasible. Additionally, the invention of new technologies, such as smartphones, requires the use of new raw materials which have not been in the production cycle previously. Thus, the production cycle must be continually supplied with new raw materials, which need to be mined.

This study has several direct outcomes. The first outcome includes two posters illustrating and explaining minerals used in daily life items. These posters are available for purchase online:

(1) the NGU poster "Nyttige mineraler og grunnstoffer (Useful minerals and elements)" utilizing minerals of the NGU collection for photographs ([https://openarchive.ngu.no/ngu](https://openarchive.ngu.no/ngu-xmlui/handle/11250/2675461?locale-attribute=en)[xmlui/handle/11250/2675461?locale-attribute=en\)](https://openarchive.ngu.no/ngu-xmlui/handle/11250/2675461?locale-attribute=en) ([Figure 23](#page-37-0)),

(2) the poster of the Natural History Museum (NHM) of Oslo "Mineraler i hverdagen (Minerals in the daily life)" ([https://webshop-uio-naturhistorisk-](https://webshop-uio-naturhistorisk-museum.trmed.no/produkt/2787/mineraler_i_hverdagen_a2)

[museum.trmed.no/produkt/2787/mineraler_i_hverdagen_a2\)](https://webshop-uio-naturhistorisk-museum.trmed.no/produkt/2787/mineraler_i_hverdagen_a2) ([Figure 24](#page-38-0)). The latter was prepared by the Mineralogy Team NORMIN with photographs by Øivind Thoresen, using minerals of the NHM collection.

The second outcome is the establishment of the Minerals-of-the-daily-life theme for the new geological NHM exhibition at Tøyen in Oslo. Results of smartphone and toothpaste analysis of this study were integrated in the content and design of the exhibition [\(Figure 25](#page-39-0) and [Figure 26\)](#page-39-1).

NYTTIGE MINERALER & GRUNNSTOFFER

Beryll

Kobber er trolig de

NORGES GEOLOGISKE UNDERSØKELSE

Kval
Det
råde
stål

Gull er et

Wyanitt, og
Kyanitt, og
som oppti
en dyp bl*i*
smykkest

FELTSPAT

Figure 23. NGU poster "Nyttige mineraler og grunnstoffer (Useful minerals and elements)".

Mineraler i hverdagen

ILMENITT fra Bamble FIND₃

Ilmenitt er et råstoff for produksjon av titanoksid.

Det er et hvitt fargepigment som tilsettes i maling,

papir, plast, kosmetikk og matvarer.

OLIVIN fra Åheim Mineraler i olivingruppen har et veldig høyt smelte-

punkt og brukes derfor i lidfaste materiale, for

eksempel som murstein for peisovner og smelteverk.

KALIFELTSPAT fra Moss Mineraler av Feltspartgruppe er svært viktige råstoffer
I glass- og keramikkproduksjon. I filsene på norske
I glass- og keramikkproduksjon. I filsene på norske
bad blir det brukt flere kilogram feltspat.

KYANITT fra Selbu AL₂SiO_e

Den viktigske egenskapen til kyanitt er at det er ildfast

Det brukes derfor til fremstilling av varmebestandig

materiale og blant annet i tennplugger, porselen og

elektriske isolatorer.

ZIRKON fra Seiland $2rSIO₄$
 $2rSIO₄$
 $2rSIO₄$
 $2rSIO₄$

stoffet zirkonium. Zirkonium i forbindelse

med aluminium er antiperspirant i deodorant

GULL fra Oppdal **EXERCT ASSESS ASSESS**

Call er et verdifullt og sjeldent edelmetall som har

fascinert fok siden stehnalderen. På grunn av sine

strømledende egenskaper blir gull mye brukt i mikro-

elektronikk, og blant annet kretskort

MOLYBDENITT fra Tokke Mo:

 $\begin{array}{lll} \mathsf{MOS}_2 & \mathsf{MOS}_2 \\ \mathsf{Molybden} & \mathsf{mtrative} & \mathsf{mrel} & \mathsf{mrel} & \mathsf{mrel} \\ \mathsf{mplybden} & \mathsf{tibettees} & \mathsf{istäl} & \mathsf{blin} & \mathsf{det} & \mathsf{mye} & \mathsf{hardere} & \mathsf{gg} & \mathsf{far} \\ \mathsf{hayere} & \mathsf{smeltepunkt}. & \mathsf{Molybdenlegeringer blir brukt i bor \mathsf{bor} & \mathsf{mtrik} & \mathsf{bor} \\ \mathsf{Mlybdenlegeringer} & \mathsf$

KOBBER fra Horten Examples the control of the control of the control of the stage of state in the stage of stage of stage of stage in the stage of the stage stage and stage i

MAGNETITT fra Larvik Fe²⁺Fe³⁺₂O₄
Magnetitt er den mest vanlige jern-
mineralet som brukes i stålproduksjon.

FLUORITT fra Modum Function of Care

Function of Care

smaller empirically and the settle mediate members will be a settlement

internation and the set of the

Fluor brukes for eksempel i tannpasta

KVARTS fra Sørli Kvarts er et viktig industrimineral med mange
bruksområder, for eksempel i glass og optiske
fibre, solcelle, stål, kosmetikk og smykkeindustri.

SINKBLENDE fra Oslo

This

Sinkblende/sfaleritt er sinksulfid som er hovedkilde

for sink og kan også inneholde grunnsforfet indium

Indium brukes mye i elektronikkomponenter og i

flytende krystall-flatskjermer (LCD).

Figure 24. NHM poster "Mineraler i hverdagen (Minerals in the daily life)".

Figure 25. Design draft of "minerals in smartphones" for the geological NHM exhibition at Tøyen, Oslo, by Atelier Brückner GmbH. On the left, minerals are shown which are needed for the manufacture of micro vibrators in smartphones, which is based on the [Figure 11.](#page-18-0) On the right, are minerals exhibited which are used for the production of other parts of smartphones. The abbreviations Ca, EP, and FP refer to label text types.

Figure 26. Design draft of "minerals in smartphones" for the geological NHM exhibition at Tøyen, Oslo, by Atelier Brückner GmbH. It illustrates the minerals used in toothpaste. The sketch on the lower left is based on [Figure 22.](#page-34-0)

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APPENDIX 1: List of specimens of the applied mineralogy exhibition at NGU.

Geological Survey of Norway
PO Box 6315, Sluppen
N-7491 Trondheim, Norway

Visitor address
Leiv Eirikssons vei 39
7040 Trondheim

(+ 47) 73 90 40 00
ngu@ngu.no
www.ngu.no/en-gb/ Tel
E-mail Web